

Commercial Refrigeration is a science of vague assumptions, based upon debatable figures, taken from inconclusive experiments, performed with instruments of problematical accuracy by persons of doubtful reliability and questionable mentality

Energy & Store  
Development Conference

2013  
E+Sd



THE VOICE OF FOOD RETAIL 

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2013

## Basic Refrigeration 101

Cooling by the  
removal of heat



The **MOVEMENT**  
of **HEAT** from a  
place where it is  
not wanted to a  
place where it is  
unobjectionable

**What is Refrigeration?**

**What is heat?**

A form of energy

**What is cold?**

Absences of heat.

**How does heat  
flow?**

High heat to low heat

**Does cold  
flow?**

NO

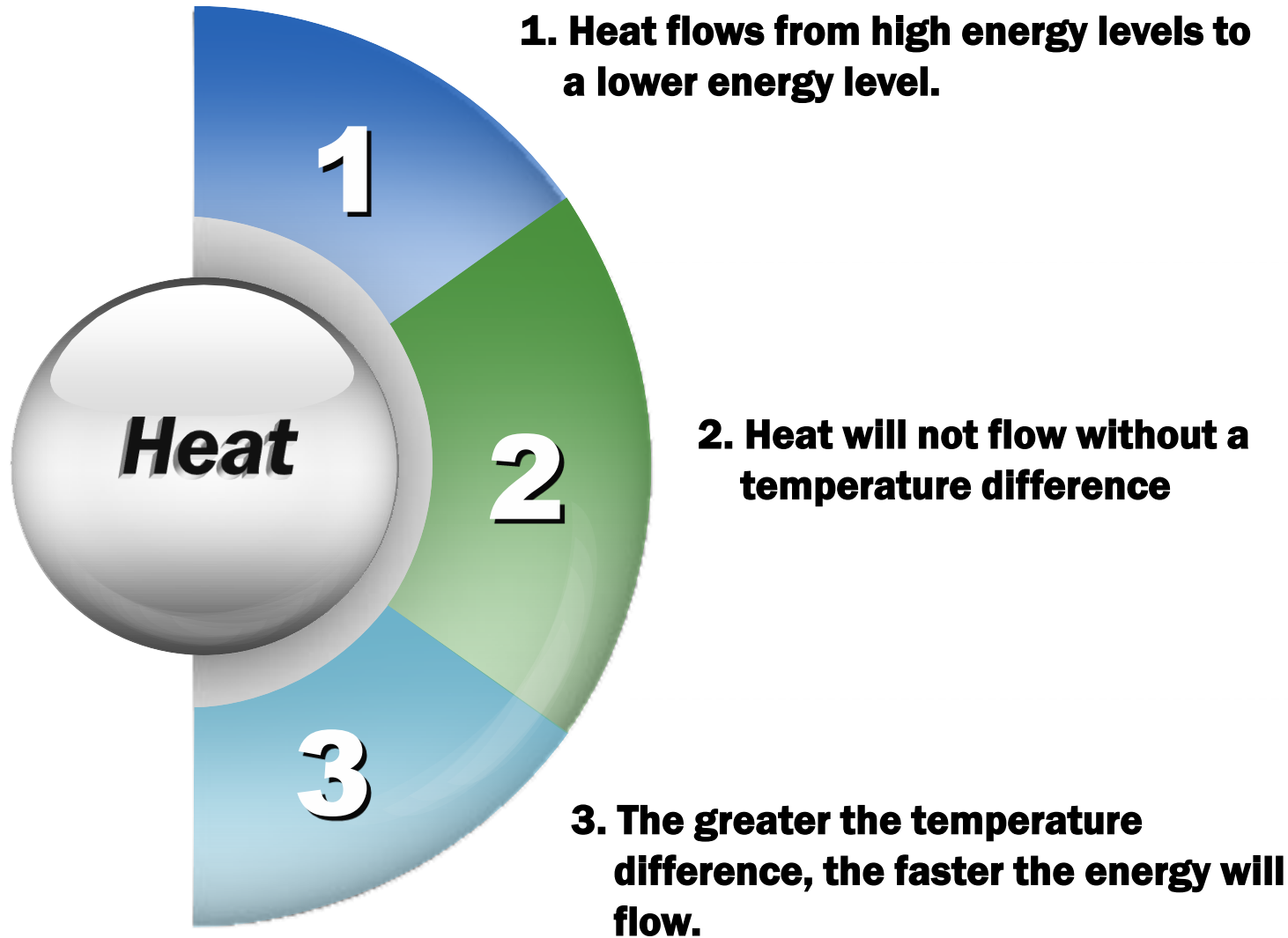
**How is heat  
measured?**

British Thermal Unit (BTU)

## **British Thermal Unit (BTU)**

**Heat is a form of energy and is measured in BTU's.**

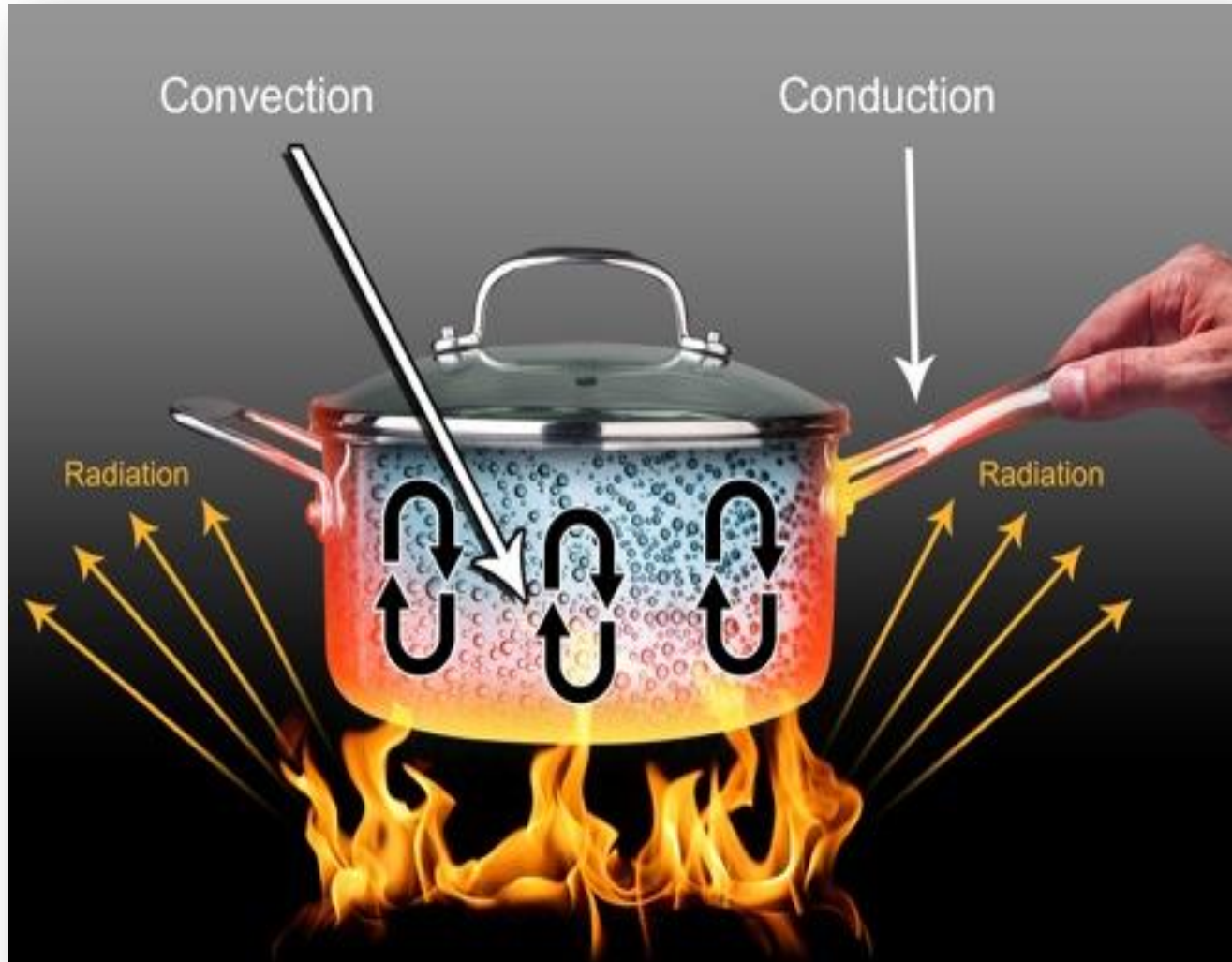
**A BTU is the quantity of heat required to raise the temperature, of one pound of water, one degree Fahrenheit.**



**Heat Flows  
Three Ways**



# Heat Flows in Three Ways



# Conduction

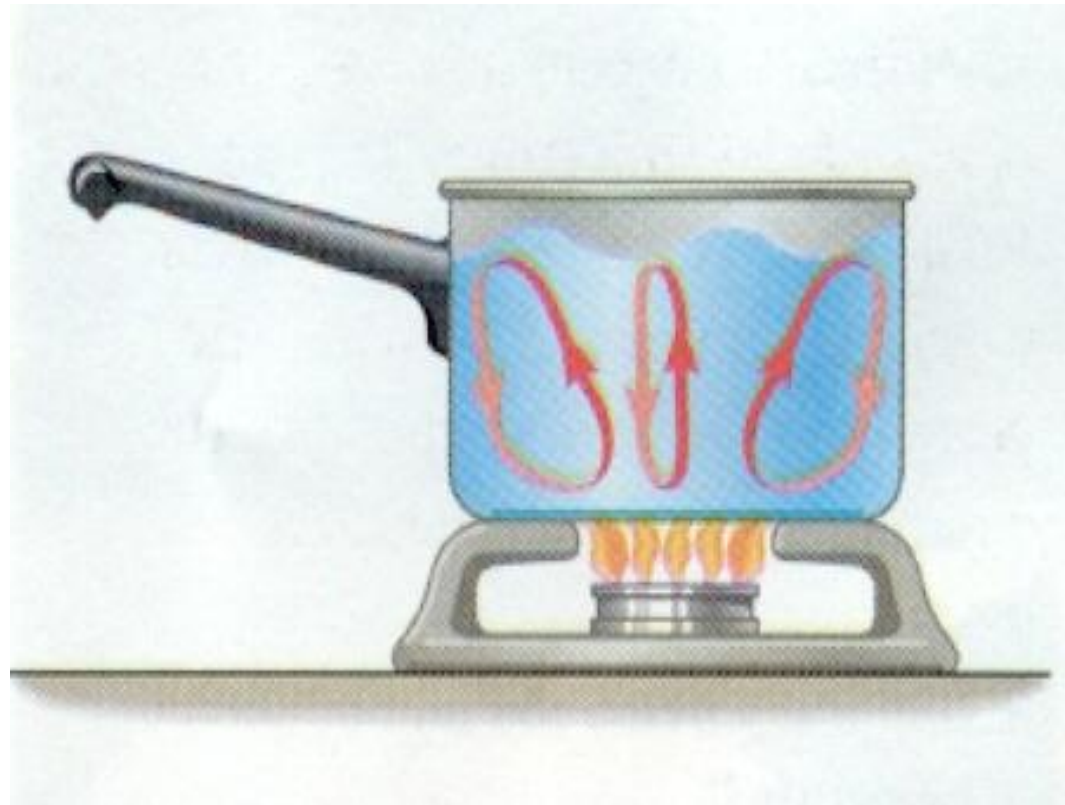
- The transfer of heat from molecule to molecule through a substance is called conduction
- Movement of heat continues until a temperature balance is achieved throughout the entire material

Heat conducts through the pipe until it is heated evenly.



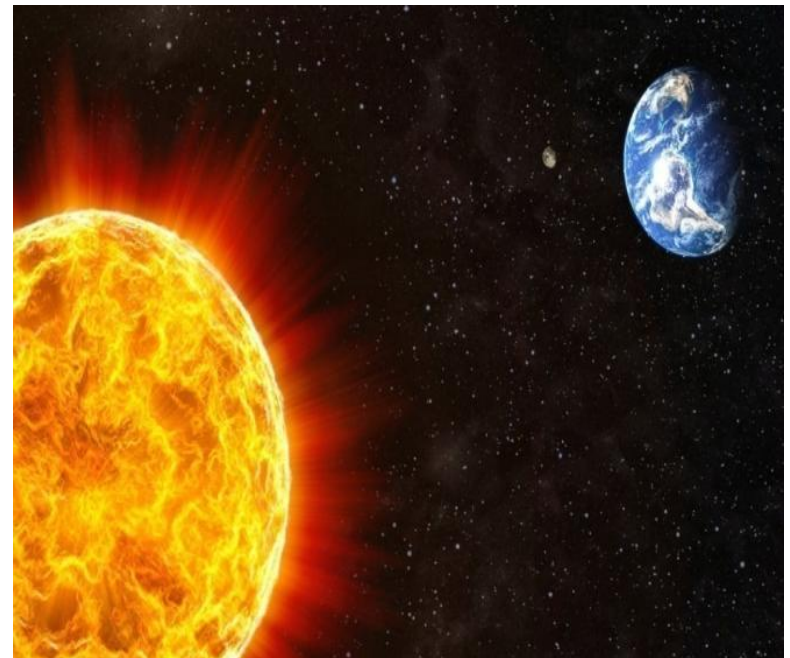
# Convection

- The transfer of heat from one place to another by movement of molecules
- Portions of a substance are set in motion, forming currents
- Air in a refrigerator moves by convection (often aided by a fan)



# Radiation

- Radiant heat is absorbed from a physically separated source without affecting the space in between
- When radiant heat is absorbed by a material, it is converted into sensible heat

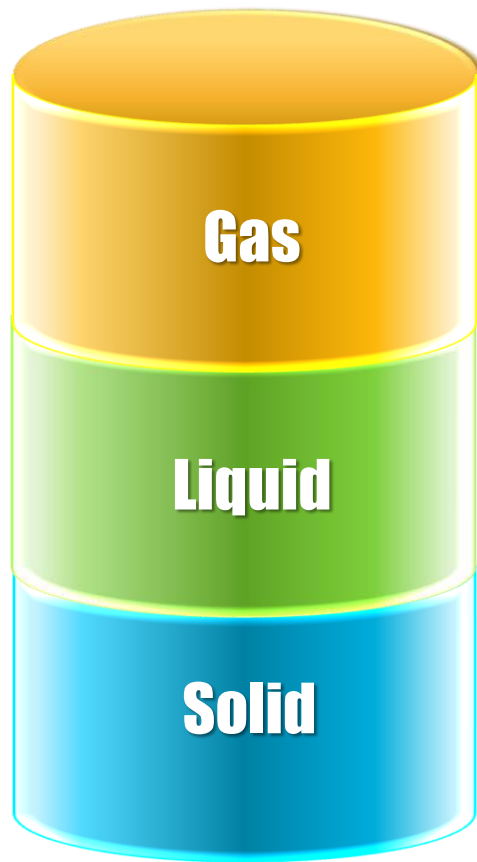


# Specific Heat

**The amount of heat (measured in BTU's) required to raise one pound of a substance one degree Fahrenheit.**

<b>Substance</b>	<b>Specific Heat (BTU/LB/Deg F)</b>	<b>Temp Rise (Deg F) (From 1 BTU Addition)</b>
Water (Liquid)	1.00	1.00
Apples	0.87	1.15
Beef, Ribs	0.67	1.49
Frog legs	0.88	1.13
Milk	0.90	1.11

# Physical State of a Substance



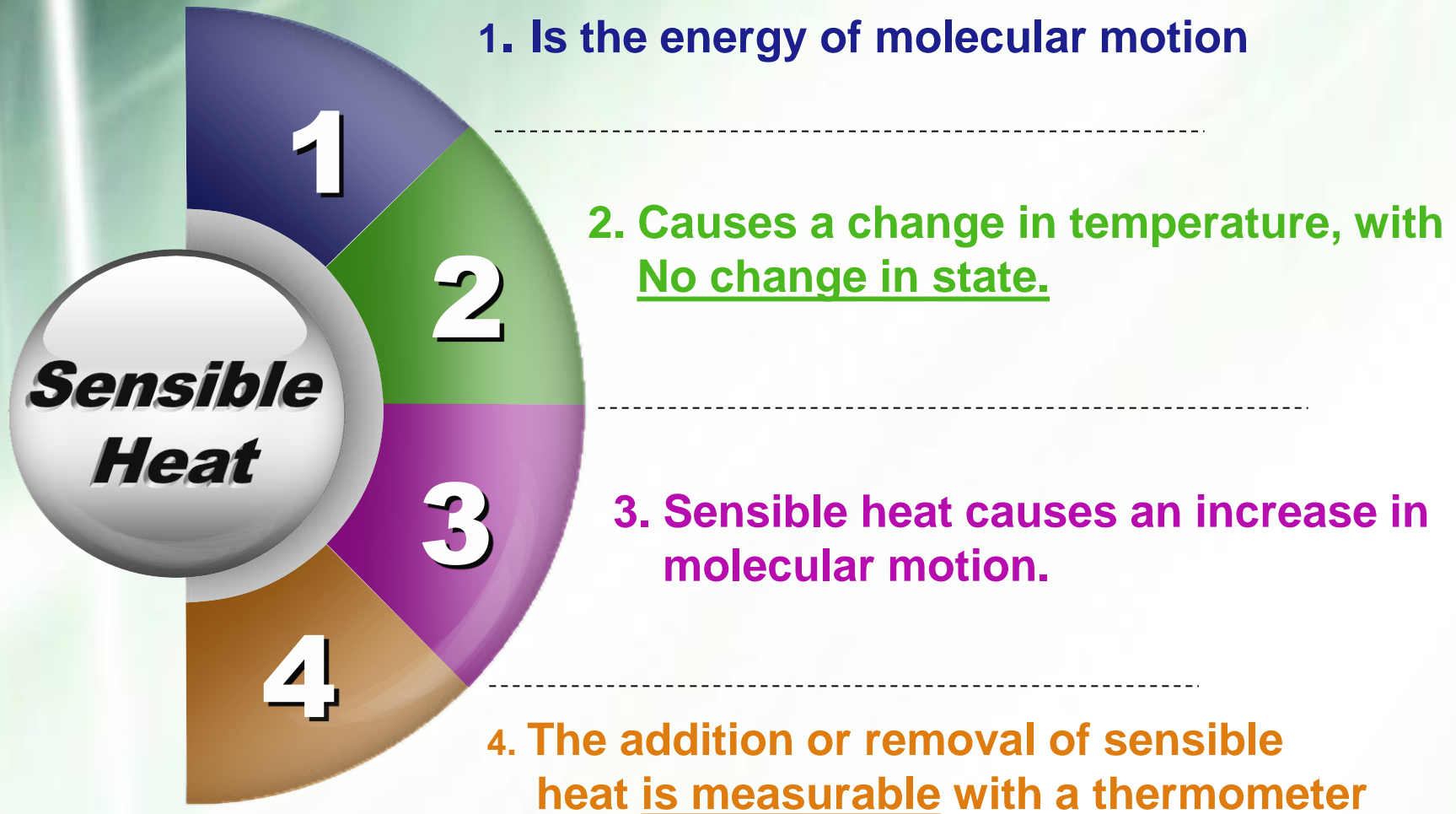
Adding heat energy in any of the three states will increase the temperature.

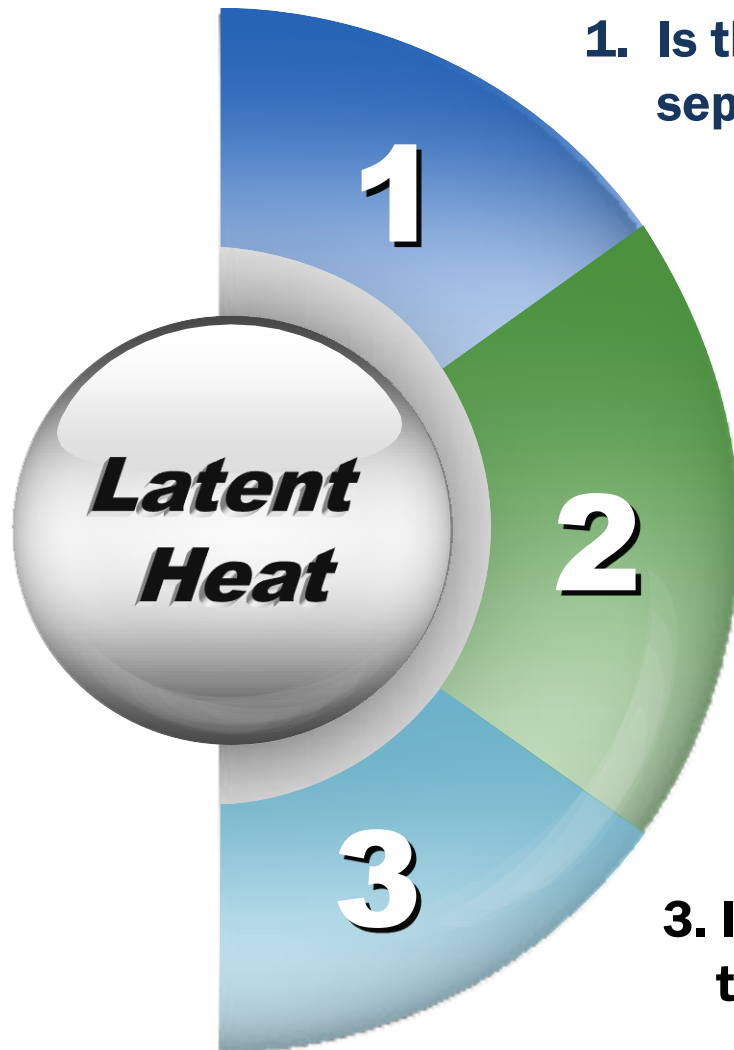
Changing the state of a substance will require a larger quantity of heat energy.

## Heat Energy

**1.**  
**Sensible**  
**Heat**

**2.**  
**Latent**  
**Heat**



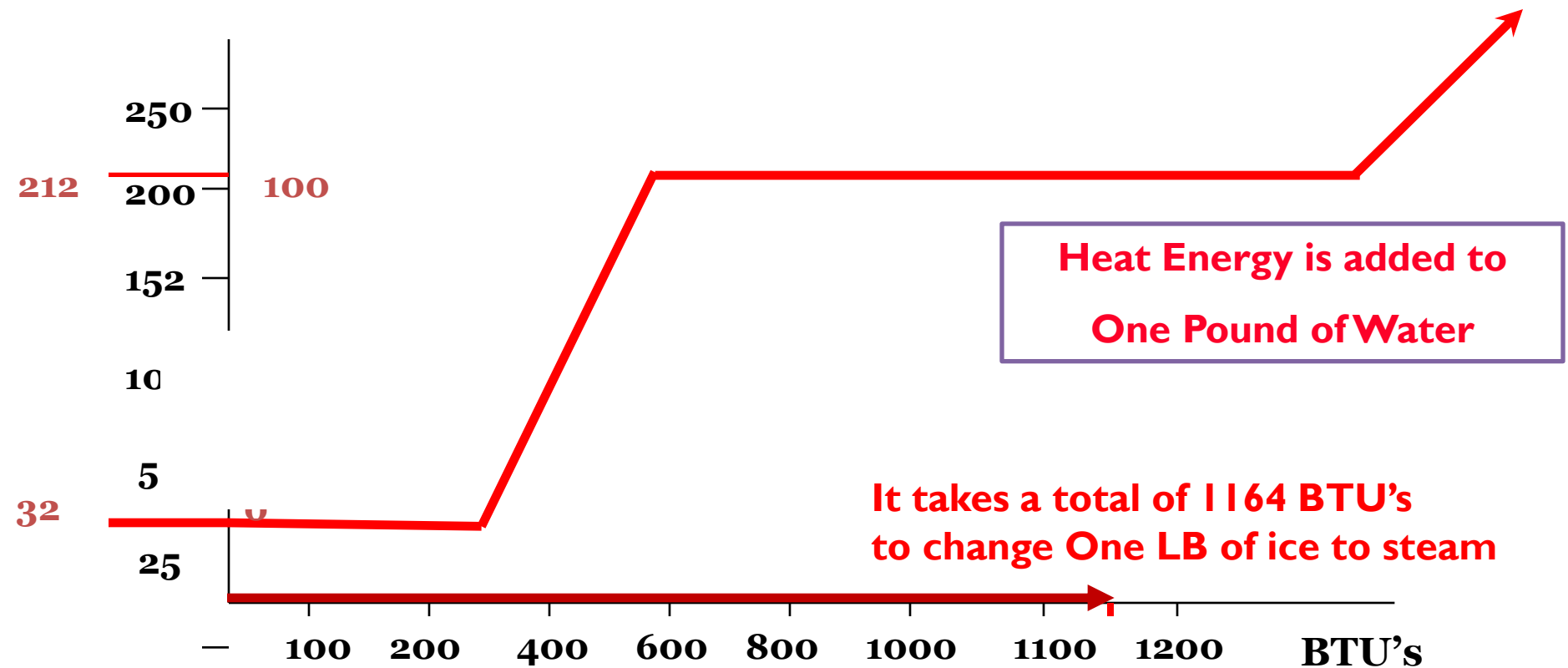


**1. Is the energy of molecular separation and arrangement.**

**2. Causes a change in state, while there is No Change In Temperature.**

**3. It can not be measured with a thermometer.**

(Deg. F.) (Deg. C.)



# Refrigeration

**Mechanical refrigeration works by changing the state of the refrigerant.**

**The majority of heat is removed from the temperature controlled space as the refrigerant absorbs heat when it changes state from a liquid to a gas in the evaporator.**

**The majority of the absorbed heat is removed from the refrigeration system in the condenser as the refrigerant changes state from a gas back to a liquid.**

# Saturation Temperature

The temperature at which a substance will change state from a liquid to a vapor or a vapor to a liquid

**Add Heat = Liquid to Vapor  
(Vaporization)**

**Remove Heat = Vapor to Liquid  
(Condensation)**

# Saturated Mixture

At a temperature equal to its boiling point or condensing point, a substance can be both vapor and liquid.



**Addition of Heat = Boiling**

**Removal Heat = Condensing**

# Sub-Cooled Liquid

**At a temperature  
below it's boiling point,  
a substance is:**

**100% Liquid, no vapor present**

**1lb. of water @ 200° F. What is the  
state of water?**

- Sub-cooled liquid**
- Sub-cooled by 12° F.**

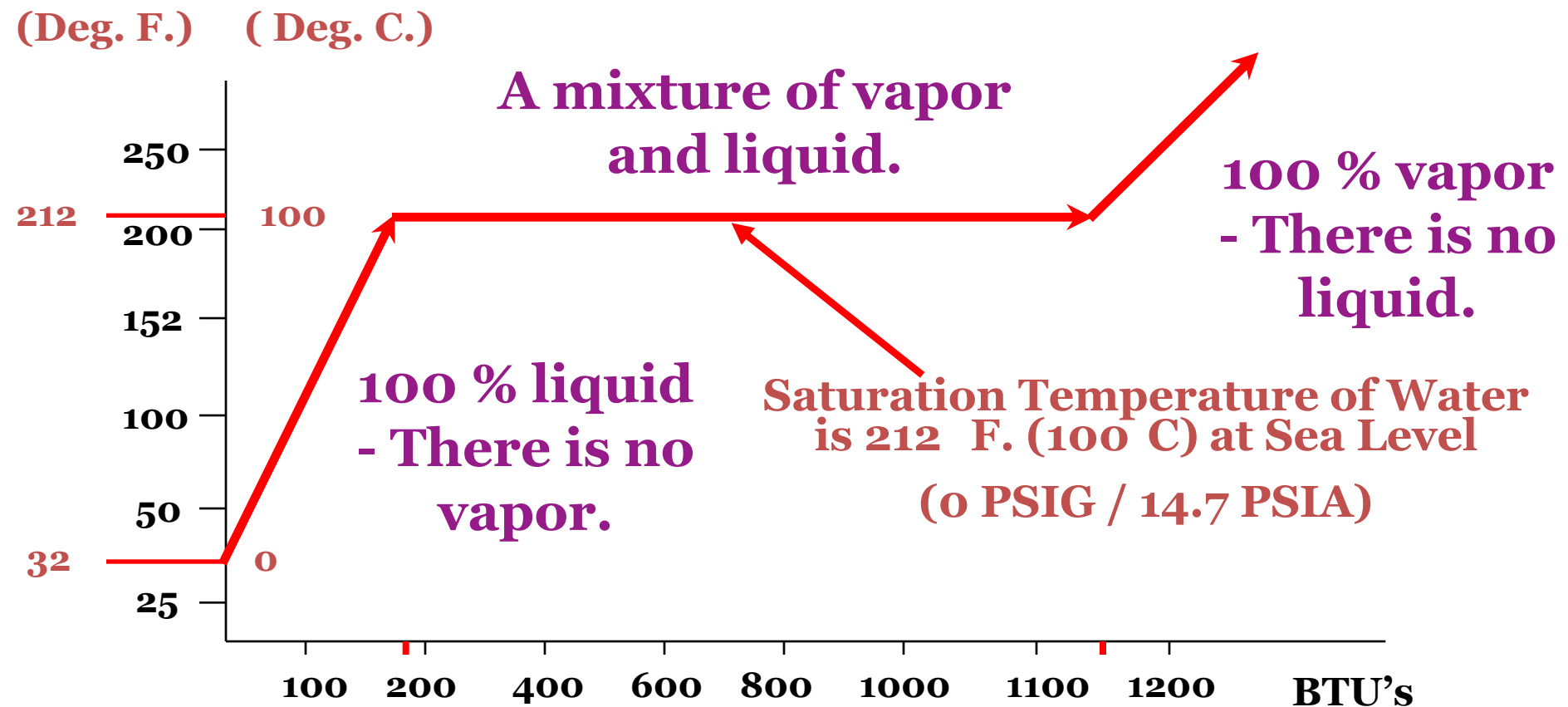
# Super Heated Vapor

**At a temperature above  
it's boiling point, a  
substance is:**

**100% vapor, no liquid present**

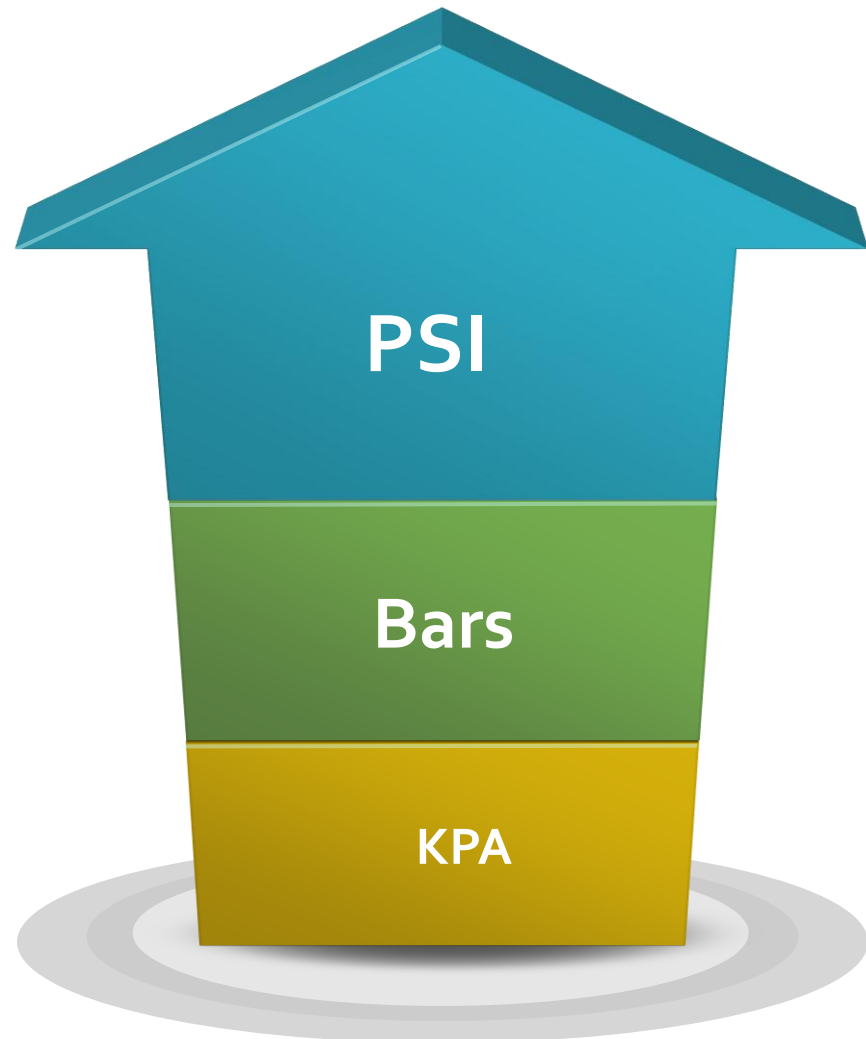
**1lb. Of water @ 220° F. What is  
the state of water?**

- Super heated vapor (steam)**
- Super heated by 8° F.**



# Pressure

**Force Per Unit of  
Area**



# Pressure Movement

**How does pressure flow?**

**High**

To

**Low**

What would happen if you take two refrigerant tanks, one with refrigerant and the other in a vacuum and connect a hose between them?

Refrigerant flows from the tank with refrigerant to the tank that is in a vacuum, until the pressures equalize.

# What is a refrigerant?

A Medium to move heat

A substance that **absorbs heat by evaporating** at low temperatures and pressures.

And **gives up heat by condensing** at high temperatures and pressures

# Pressure as it relates to Temperature


Temp. °F	REFRIGERANT					
	R-507 (AZ-50)	R-404A (HP-62)	R-410A (AZ-20)	R-22	R-744 (CO <sub>2</sub> )	R-717 (NH <sub>3</sub> )
104	256.2	250.6	337.1	207.7		210.9
105	259.9	254.2	341.9	210.8		214.4
106	263.5	257.8	346.7	213.8		217.9
107	267.2	261.4	351.5	216.9		221.5
108	271.0	265.1	356.5	220.0		225.1
109	274.8	268.8	361.4	223.2		228.8
110	278.6	272.5	366.4	226.4		232.5
111	282.4	276.3	371.5	229.6		236.2
112	286.3	280.1	376.6	232.8		240.0
113	290.3	284.0	381.8	236.1		243.9
114	294.2	287.9	387.0	239.4		247.7
115	298.3	291.8	392.3	242.8		251.6
116	302.3	295.8	397.6	246.1		255.6
117	306.4	299.8	403.0	249.5		259.6
118	310.5	303.8	408.4	253.0		263.7

Temp. °F	REFRIGERANT					
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17	54.3	51.5	73.2	39.9	388.0	30.4
18	55.6	52.7	74.9	40.9	394.3	31.4
19	56.9	54.0	76.6	42.0	400.7	32.4
20	58.2	55.3	78.4	43.1	407.2	33.5
21	59.6	56.6	80.1	44.2	413.8	34.6
22	61.0	58.0	81.9	45.3	420.4	35.6
23	62.3	59.3	83.7	46.5	427.1	36.8
24	63.8	60.7	85.5	47.6	433.8	37.9
25	65.2	62.1	87.4	48.8	440.7	39.0
26	66.7	63.5	89.2	50.0	447.6	40.2
27	68.1	64.9	91.1	51.2	454.6	41.4
28	69.6	66.4	93.1	52.4	461.7	42.6
29	71.1	67.8	95.0	53.7	468.8	43.8
30	72.7	69.3	97.0	55.0	476.1	45.0
31	74.2	70.8	99.0	56.2	483.4	46.3


 Past


 Majority  
of New  
Systems

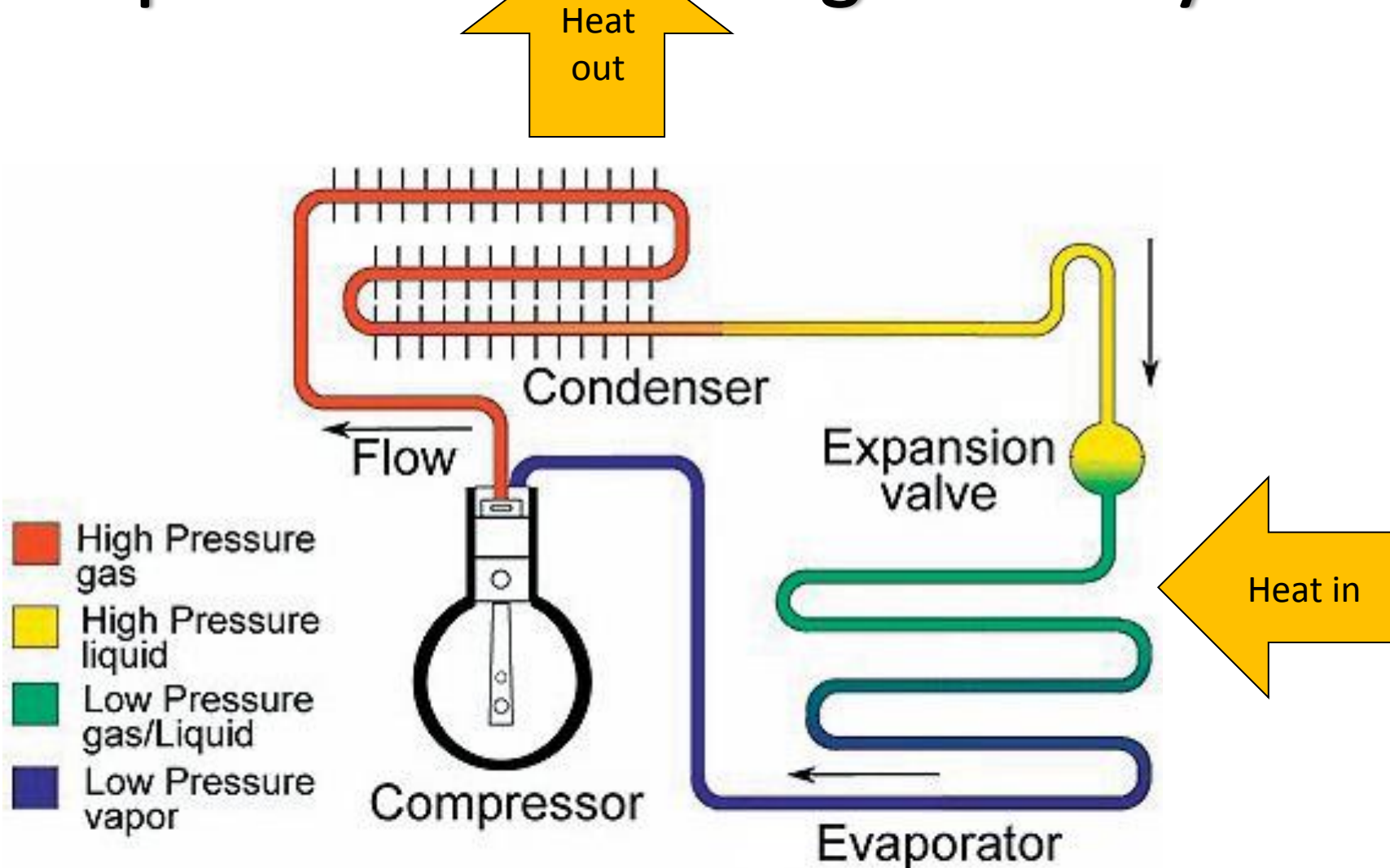

 Other  
Retrofit  
Fluids

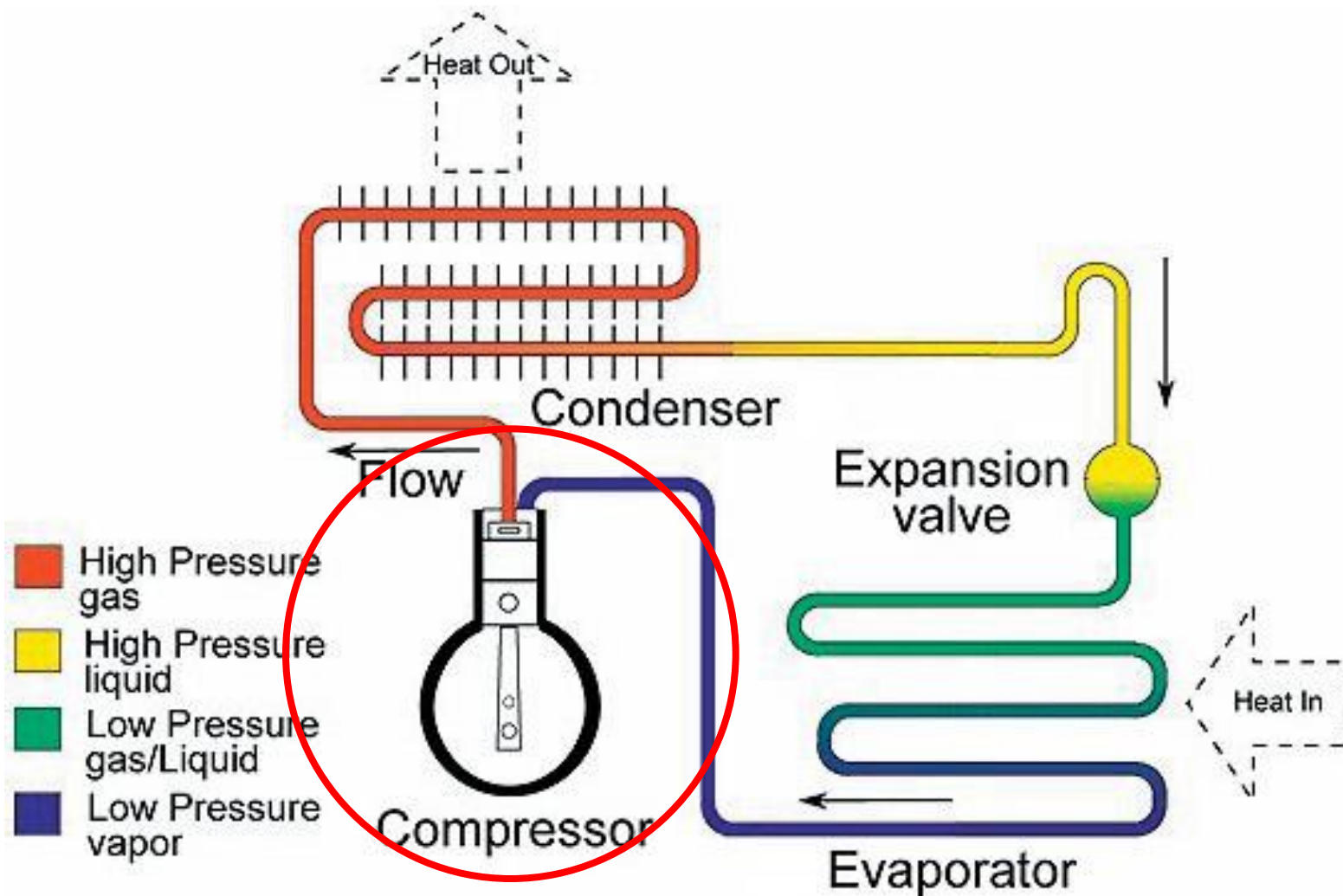

 Higher  
Pressure

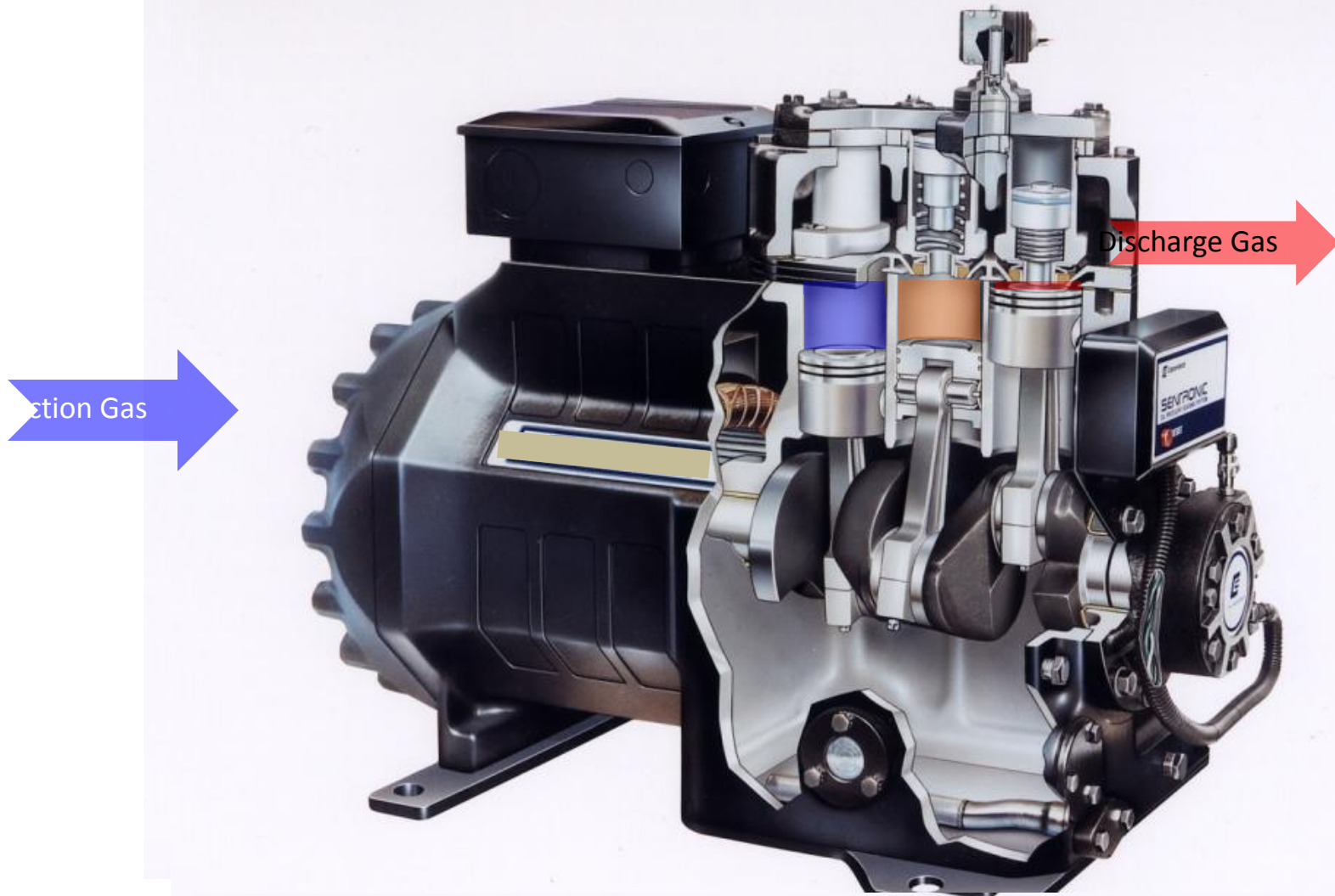

 Natural  
Fluids

Refrigerant	Other Names/Components	ODP
R-22	Freon-22 (HCFC)	0.055
R-404A	R-125/143a/134a, HP-62	0
R-507	R-125/143a, AZ-50	0
R-407A	R-32/125/134a, Klea-60	0
R-407C / 407F	R-32/125/134a, Klea-66	0
R-422A	R-125/134a/600a, Isceon MO-70	0
R-422D	R-125/134a/600a, Isceon MO-29	0
R-410A	R-32/125, AZ-20, Puron	0
R-744	CO <sub>2</sub>	0
R-717	Ammonia, NH <sub>3</sub>	0
R-290	Propane (HC)	0
R-600a	Isobutane (HC)	0
R-600	Butane (HC)	0

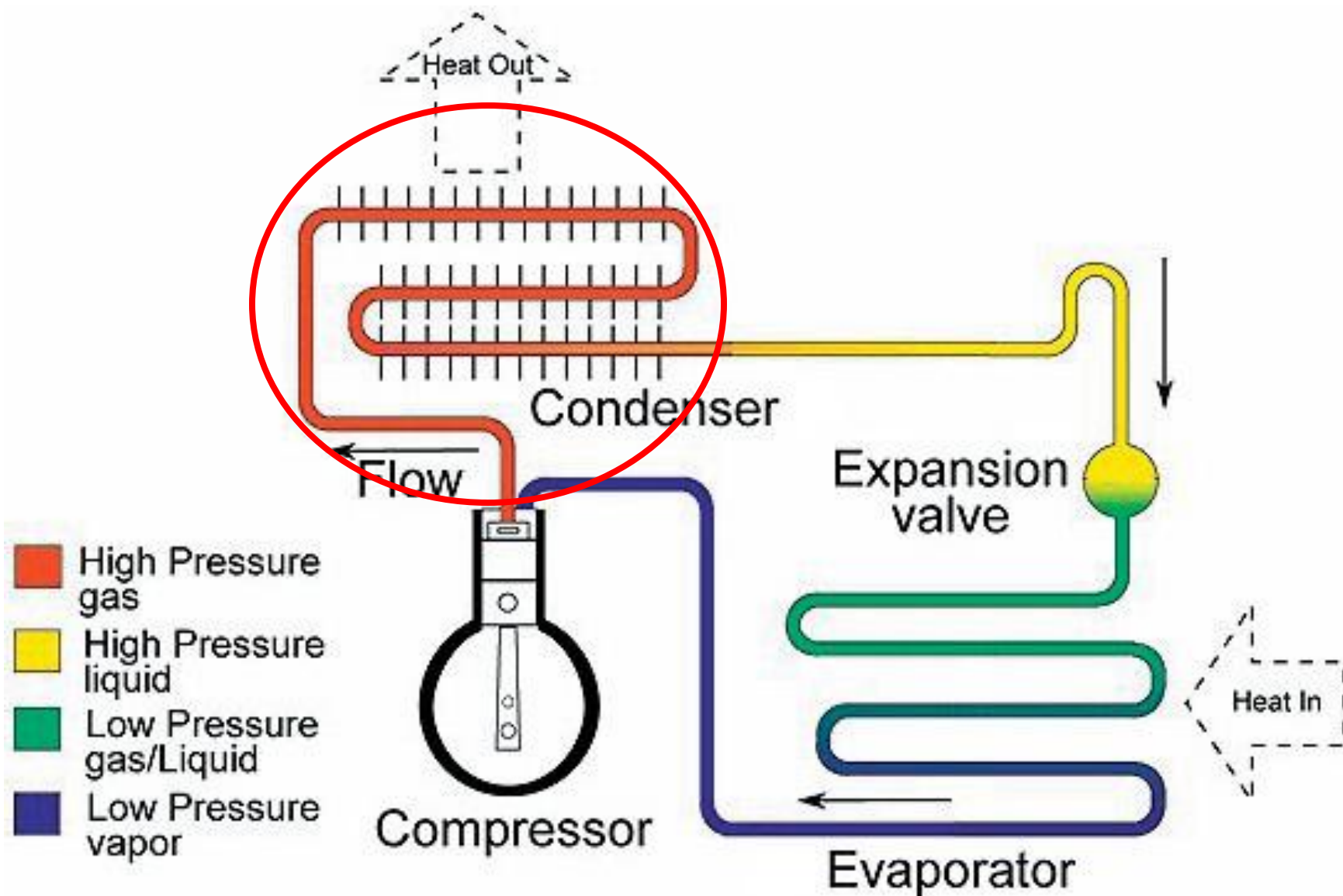
# Components of the refrigeration cycle

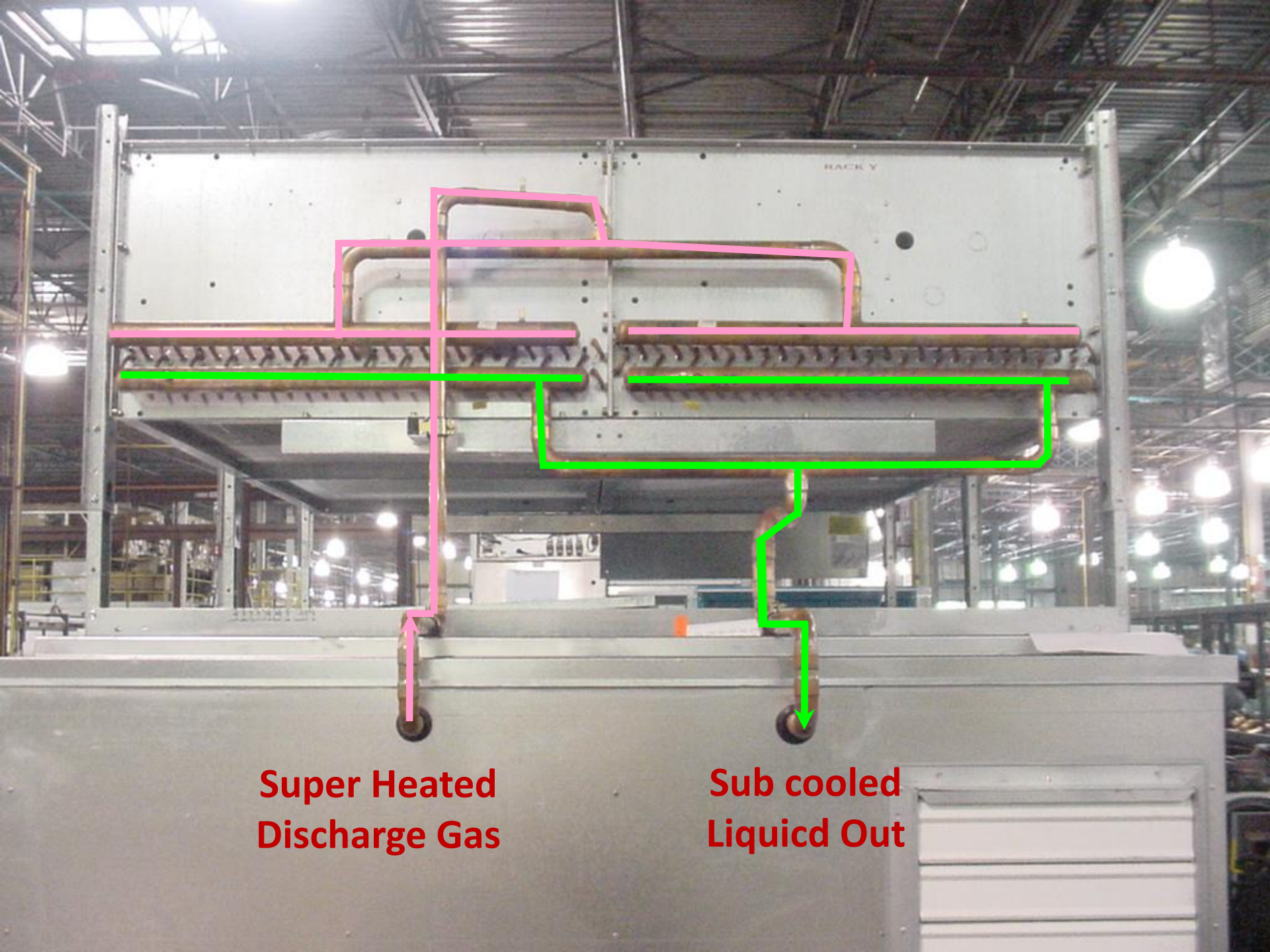










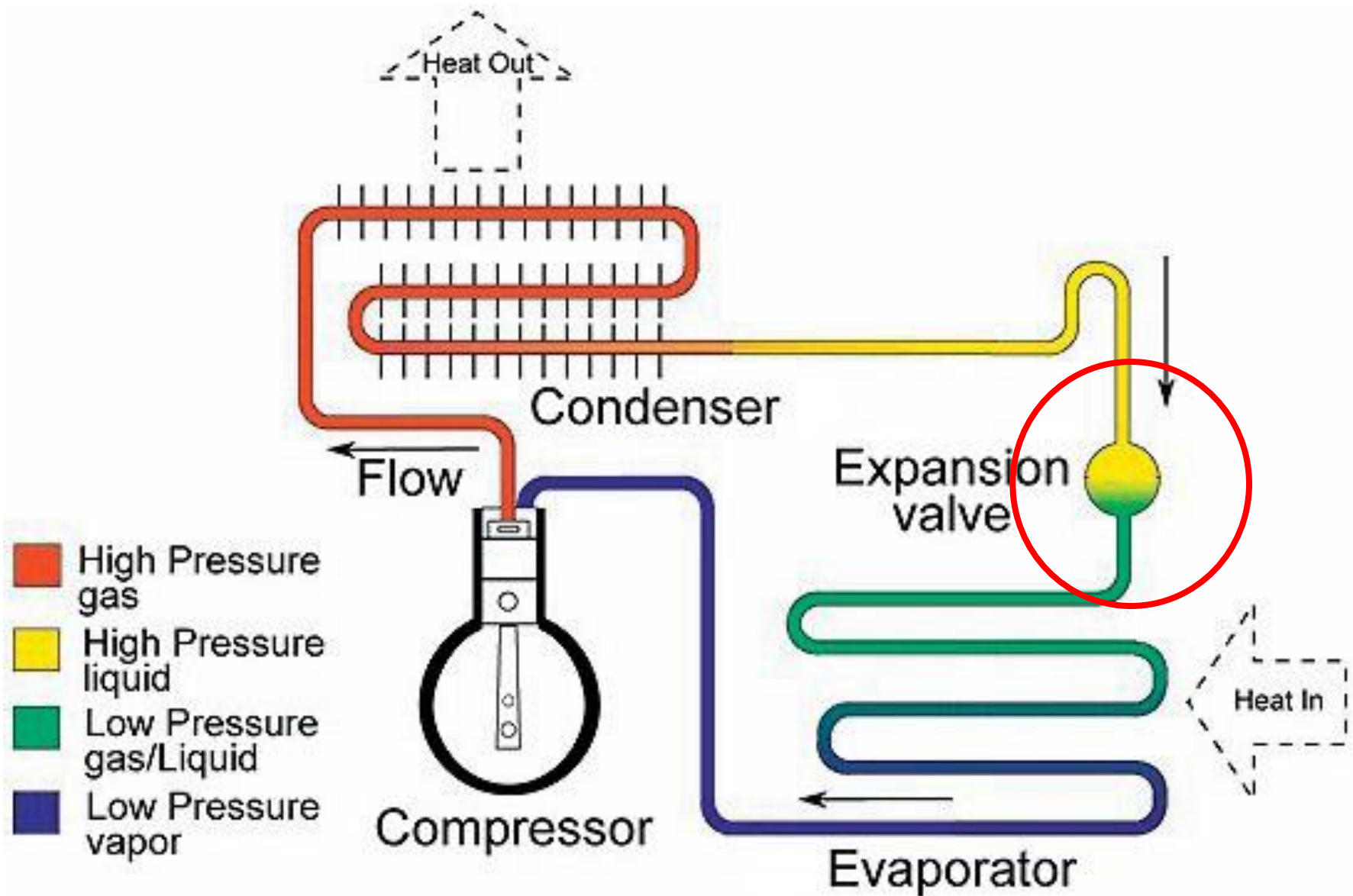


**Super Heated  
Discharge Gas**

**Sub cooled  
Liquid Out**









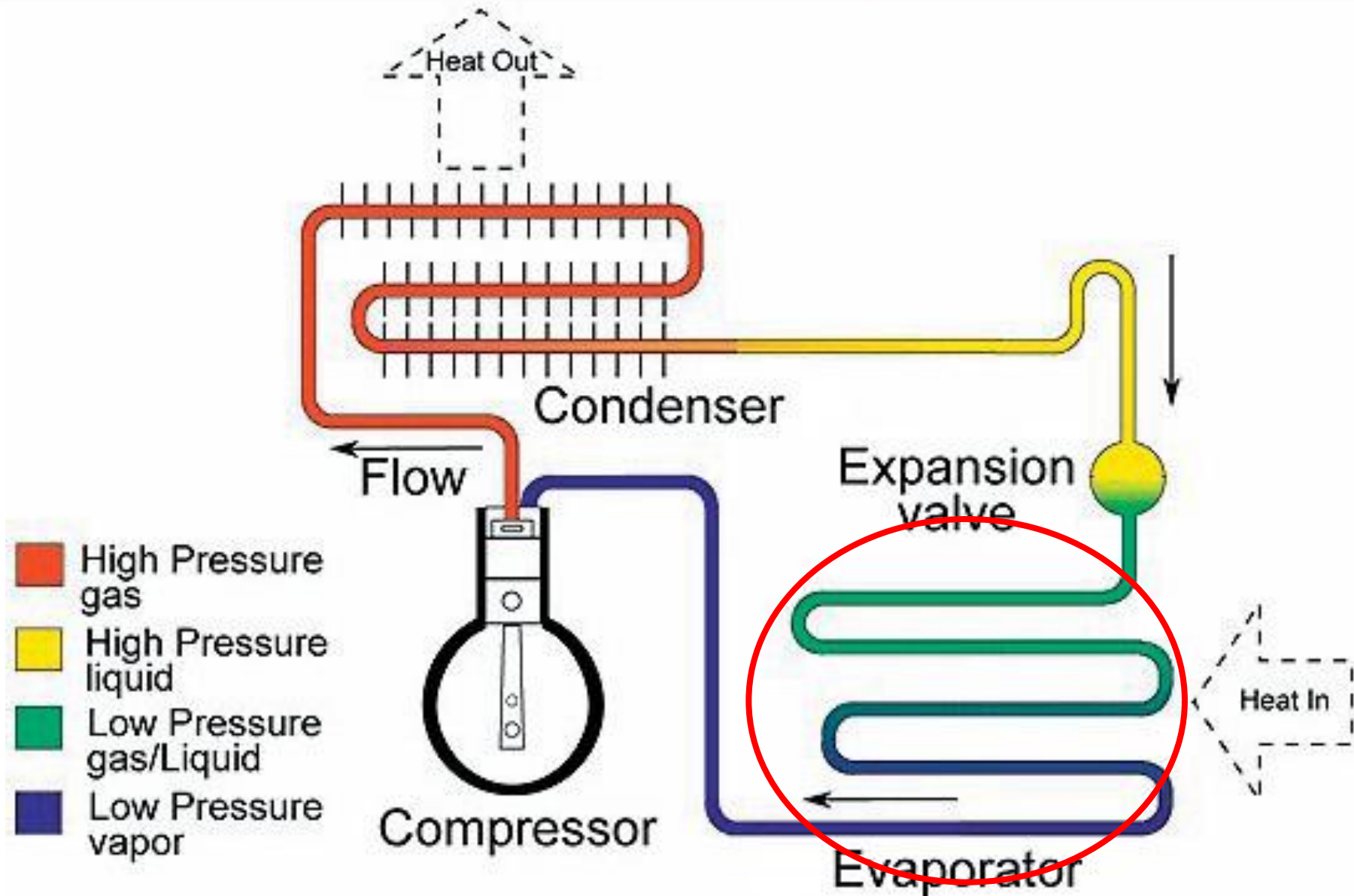
# Pressure as it relates to Temperature

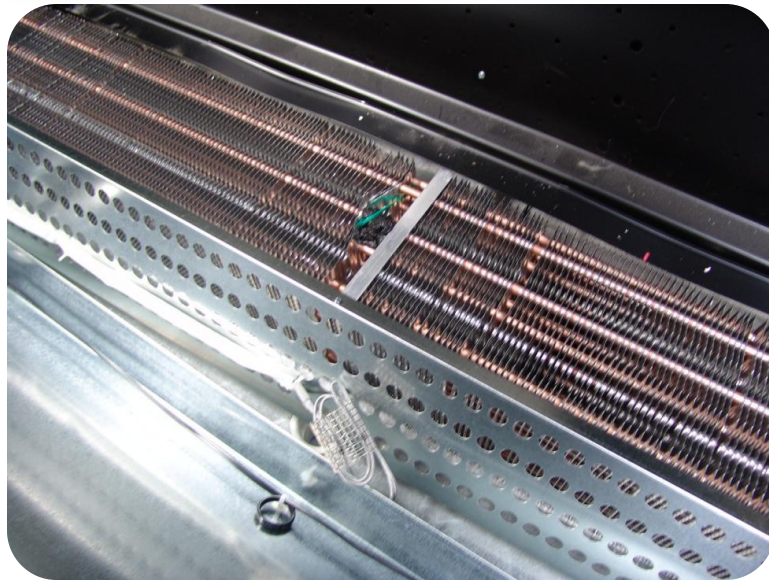
Pressure Entering TXV

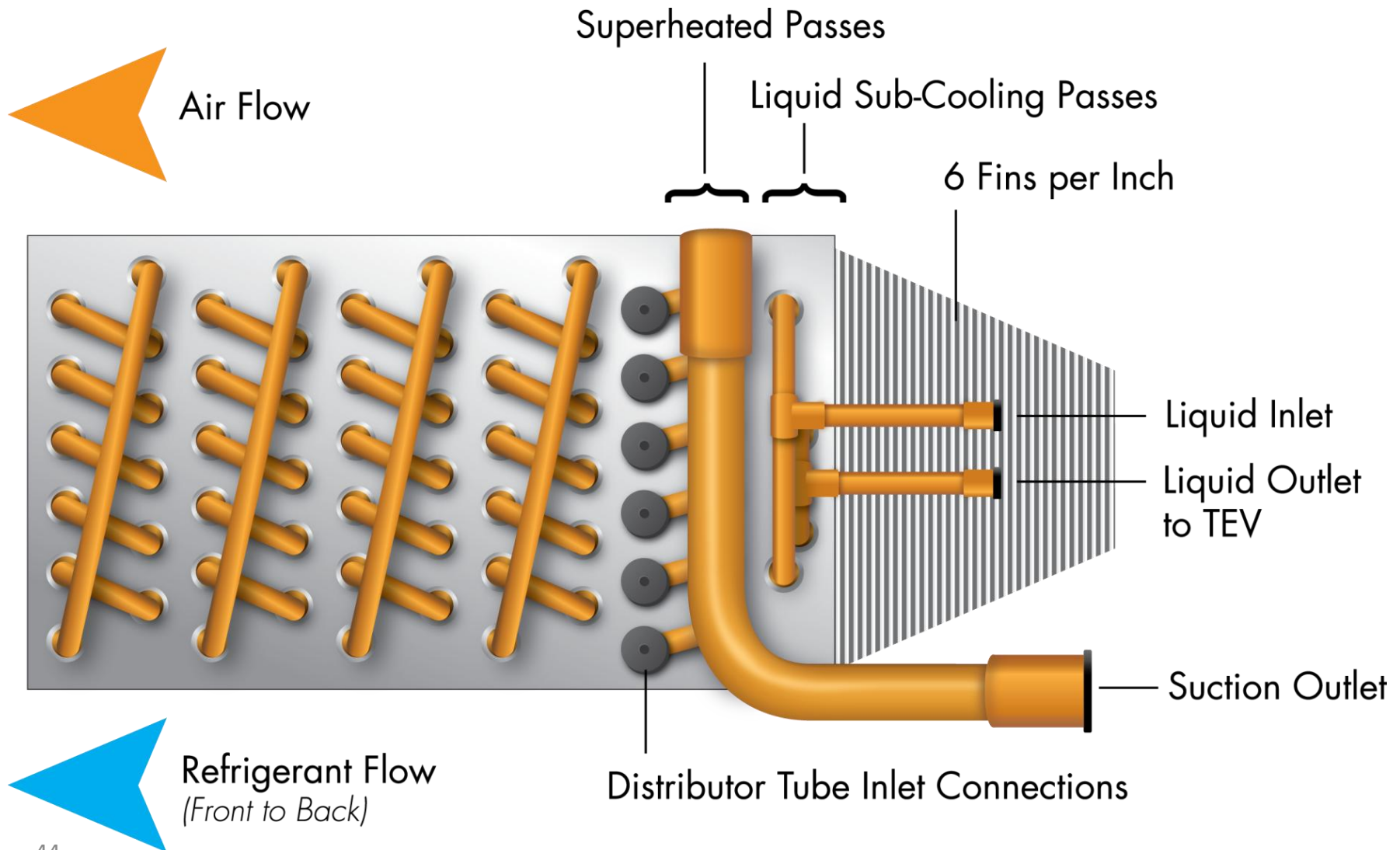
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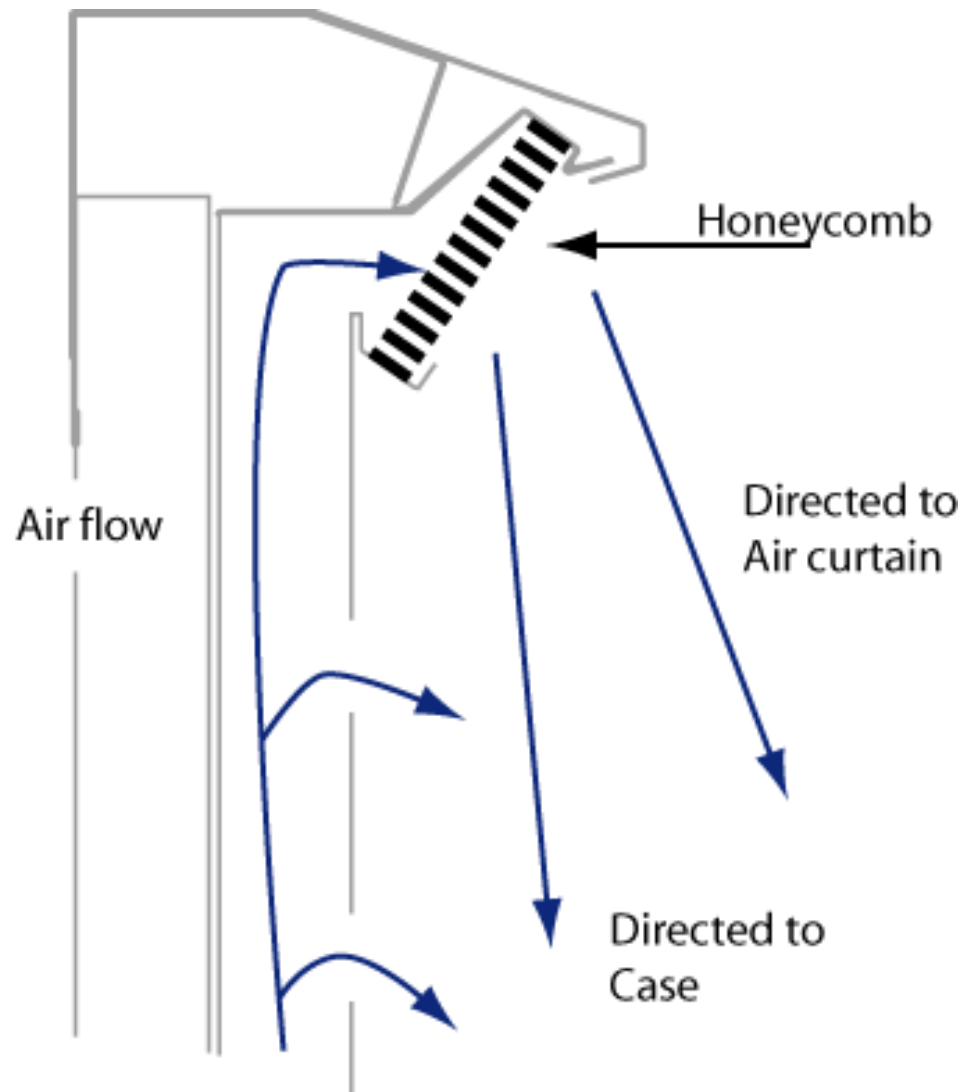
Pressure Leaving TXV

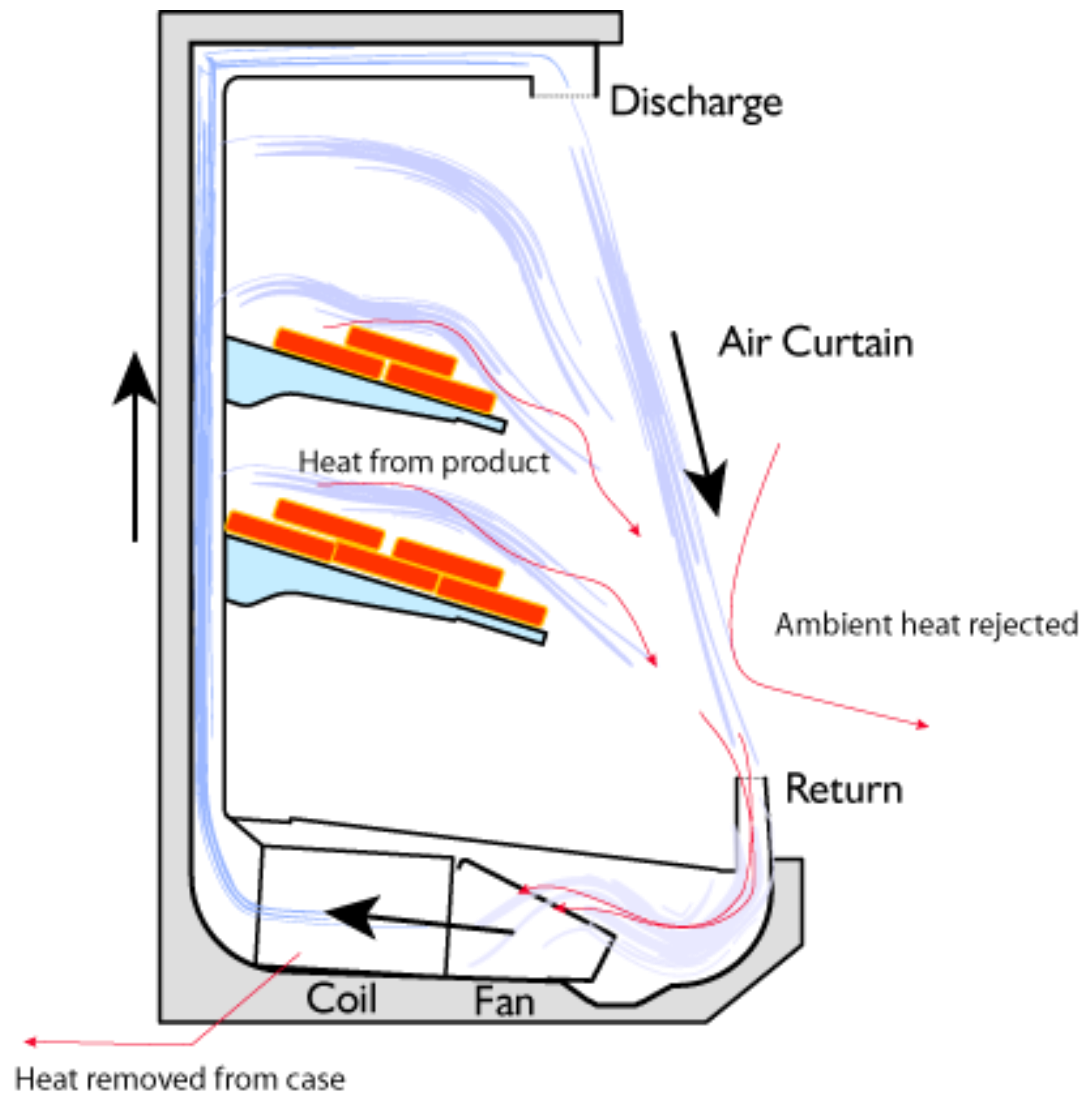
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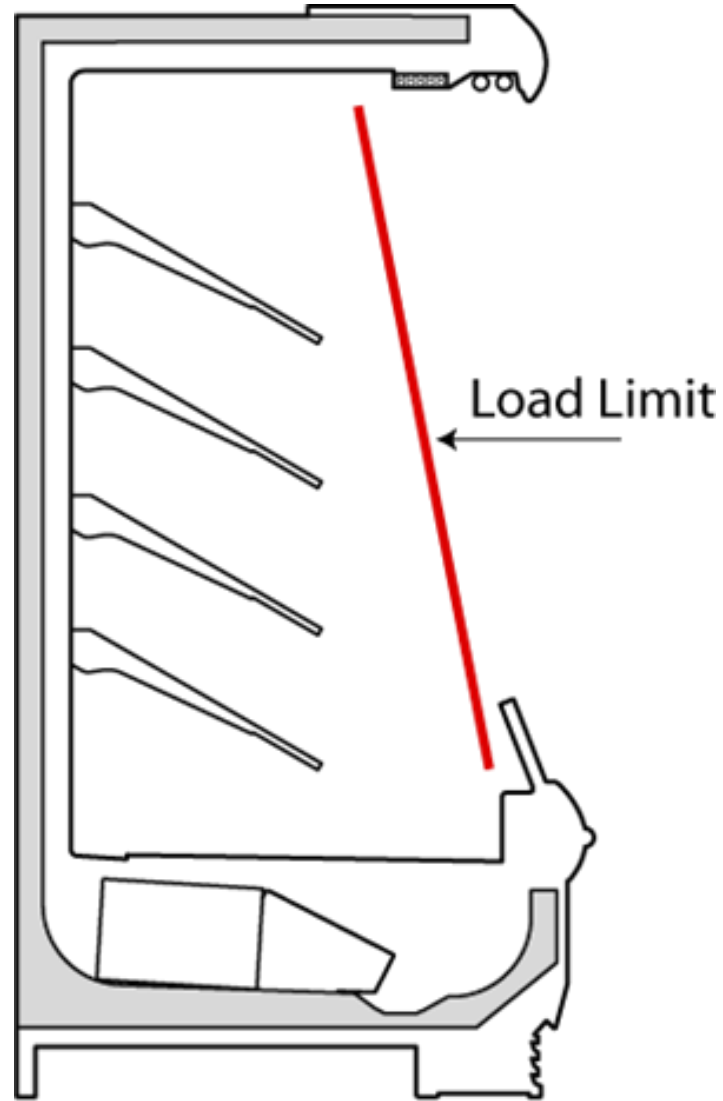


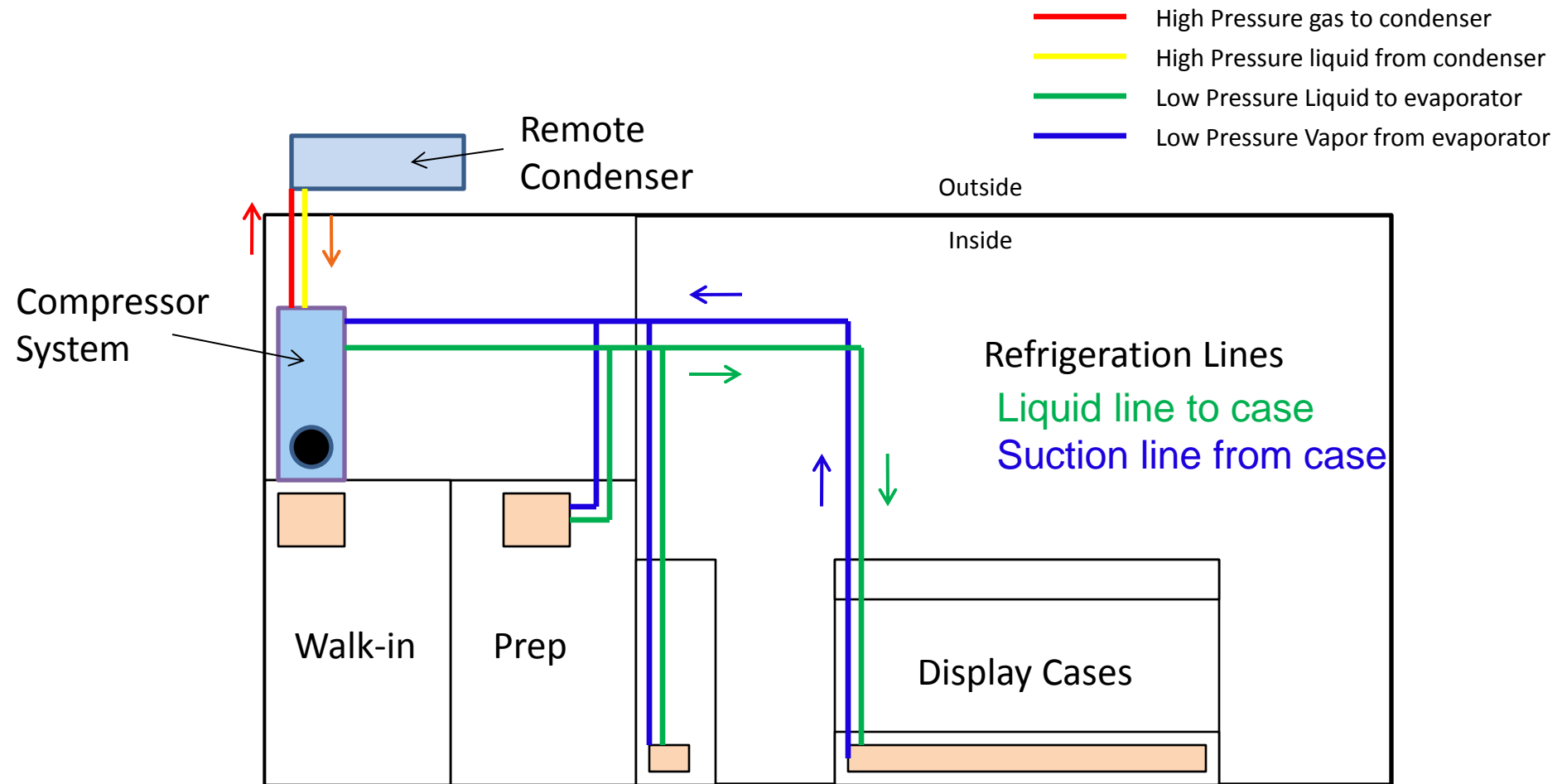












**How the components tie together**

# Questions